



Towards the Discrimination between Natural and **Synthetic Pigments: The Case of Ultramarine**

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Introduction

Ultramarine has been for centuries one of the most highly prized pigments of all traditional artists' materials, due to its durability, excellent color, and its intrinsic value. For the production of Ultramarine, the rare semiprecious stone Lapis Lazuli is used. The production of a synthetic version by Guimet in 1828, which was obtained from the calcination of a mixture of metakaolin, sulfur, sodium carbonate, and a reducing agent, followed by an oxidation step, introduced an important change in artists' habits, in that a less expensive pigment was available for their palettes¹. The verification of the natural or synthetic origin of the pigment is particularly important in a project, e.g. in authentication cases.

Materials and Methods

Νο	Pigment	Origin	Composition
1	Ultramarine Blue, very dark	4500, Kremer	Synthetic mineral pigment
2	Ultramarine blue, light	4508, Kremer	Synthetic mineral pigment
3	Ultramarine Blue, dark	45010, Kremer	Synthetic mineral pigment
4	Ultramarine Blue, Belgian	99750, Kremer	Synthetic mineral pigment (Historical stock)
5	Ultramarine Blue, reddish	45020, Kremer	Synthetic mineral pigment
6	Ultramarine Blue, greenish light	45040, Kremer	Synthetic mineral pigment
7	Ultramarine Blue, greenish extra	45030, Kremer	Synthetic mineral pigment
8	Lapis Lazuli, good quality	1052, Kremer	Natural ultramarine



- transform infrared (FTIR) Fourier spectroscopy: Spectrum 1000 Perkin Elmer, mid-IR region (4000-400 cm⁻¹), transmittance mode, 32 scans, 4 cm⁻¹ resolution, 1% KBr pellets.
- X-ray Diffraction (XRD): Rigaku Ultima, Bragg Brentano, CuKa, 40 KV, 30 mA, range (2θ): 5-90°, step size: 0.05°, count step: 2.0 s/step
- X-ray Photoelectron spectroscopy (XPS): Kratos Analytical Axis UltraDLD (Shimadzu).
- SEM-EDS Microscopy: JEOL JMS-840A with Oxford ISIS 300 detector, 20 kV accelerating voltage

Di

Ka

Ka

CC

H

70

Ka

Ka

UV-Vis Spectrophotometry: Lambda18 Lapis Lazuli from Chile 9 1056038, Kremer Natural ultramarine Diffraction spectroscope (Perkin Elmer). X-ray (XRD): Ultima Plus (Rigaku). Natural ultramarine 10 Lapis Lazuli, bright pure blue 10550, Kremer from South America

Results



Fig 1. The slight differences between the spectra of the natural and synthetic ultramarine are related to the presence of reagent residues, such as kaolinite in the synthetic pigment, whereas in the natural pigment calcite vibrations also occur. In general, all spectra are characterized by the fundamental vibrations of Al, and Si-O₄².



Fig. 2 a) X-Ray diffractograms of the pigments. 2 b) The percentages of the crystalline phases of the pigments. L:Lazurite, H:Haüyne, Ka: Kaolinite, CC: Calcite, Di: Diopside, P:Phlogopite, S:Sodalite, W: Wollastonite.

XPS





SEM-EDS



Artificial (Fig. 5, sample 1) and natural (Fig. 6, sample 8) pigments mainly contain sodium, aluminum, silicon and sulfur, as proved by SEM-EDS microanalyses⁴. Differences in grain morphology are noted.

UV-Vis

Pigment	\mathbf{L}^{*}	$\boldsymbol{\alpha}^*$	b*	С	\mathbf{h}_0	Colour

Fig. 3a), Fig. 4a) provide the identification and assignment of the binding energies of all the elements of a natural ultramarine (Sample 8) and an artificial one (Sample 1) on wide-scan XPS spectra. Fig. 3b), Fig. 4b) The atomic concentrations of sulfur species calculated from the areas of doublets S $2p_{3/2}-2p_{1/2}$. $S_x^{2^-}-p_{3/2}$ polysulfide (x > 2) and $SO_4^{2^-}-sulfate$. Fig. 3 c), Fig 4 c) The spectrum of O 1s has been fitted with two peaks, Si₂O₆ (Diopside phase)- $SO_4^{2^-}$ in the natural pigment and SiO₄ (Haüyne and Lazurite phases) in the artificial one.⁵

Conclusions

In this research work, ten samples of natural and synthetic ultramarine pigments were used to investigate the possibility of their discrimination using characterization methods. The differences between natural and synthetic ultramarine pigment are distinguished by the crystalline phases contained. Natural pigments show a greater variety of crystalline phases (Diopside, Phlogopite, Wollastonite, Sodalite) containing the elements mainly of e.g. Ca, Mg, F, while all samples show as main phases of lazurite and haüyne. This differentiation was shown through XRD with phase identification, through XPS from identification of the sulfur radical anions doublets with the appropriate assignment of compounds. Through SEM-EDS and UV-Vis characterization techniques, the percentages of elements in the samples and the absorbances corresponding to the blue colour were confirmed by colorimetric study for each sample, respectively.

1	32,1	4,8	-69,3	69,5	274,0	
5	33,2	2,8	-66,8	66,9	272,4	
4	33,9	1,2	-71,0	71,0	271,0	
7	34,8	0,8	-77,3	77,3	270,6	
6	37,6	-3,1	-73,0	73,1	267,6	
3	38,1	2,1	-66,3	66,4	271,8	
10	47,7	-3,9	-42,5	42,7	264,8	
9	48,7	-3,1	-33,9	34,0	264,8	
8	55,7	-5,8	-33,5	34,0	260,1	
2	59,7	-4,4	-47,0	47,2	264,7	



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